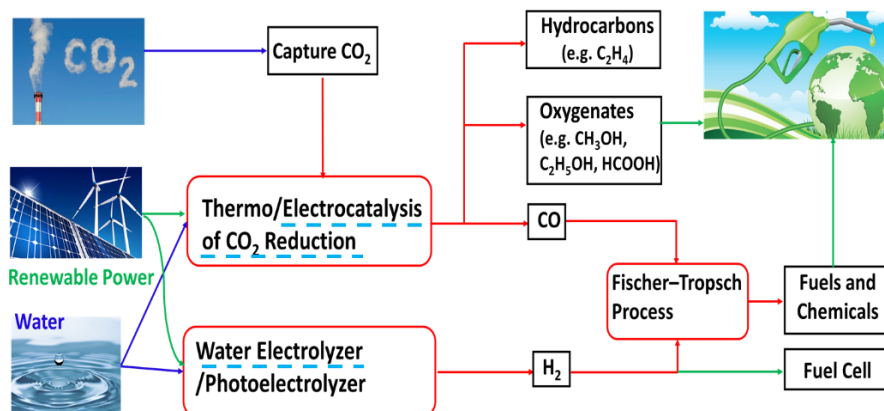


**Protégé Topic:** Electrocatalytic Conversion of Carbon Dioxide into Ethanol

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The electrochemical conversion of carbon dioxide ( $\text{CO}_2$ ) into liquid fuels is a technology to recycle carbon while also storing intermittent renewable energy (e.g. wind and solar) into chemical energy (**Figure 1**). The advancement of this technology is currently limited by the lack of **1)** efficient and stable catalysts, and **2)** operative electrode architecture for solid-state electrochemical cell employment. The traditional metal catalysts require high overpotential (low energy efficiency) for the electrocatalytic  $\text{CO}_2$  reduction reaction ( $\text{eCO}_2\text{RR}$ ) due to the “linear scaling relationship” between reaction intermediates adsorption energy. More importantly they are deficient in C-C coupling to produce ethanol ( $\text{C}_2\text{H}_5\text{OH}$ ). Additionally, they face a serious durability issue. The current electrochemical cell for  $\text{eCO}_2\text{RR}$  includes a buffer layer through which liquid neutral or alkaline electrolyte flows.<sup>1</sup> The involvement of liquid electrolyte not only results in large Ohmic loss but also causes extra cost in modular assembly.

## Scope of Research Fuels from $\text{CO}_2$ and $\text{H}_2\text{O}$



**Figure 1.** The catalytic approach for conversion of  $\text{CO}_2$  into fuels and chemicals.

This project dedicates to achieve direct  $\text{eCO}_2\text{RR}$  to produce  $\text{C}_2\text{H}_5\text{OH}$  in a solid-state electrolyzer by realizing the following objectives: **1)** develop carbon materials based catalysts through nanoscale design of topological structure; **2)** maximize the triple-phase interface boundary in the catalyst layer by percolating theory; **3)** design and manufacture a continuous flow solid-state electrolyze prototype by 3D printing. The refined electrochemical system targets  $\text{eCO}_2\text{RR}$  into  $\text{C}_2\text{H}_5\text{OH}$  with a Faradaic efficiency (FE) of 90%, energy efficiency of 50%, current density  $600 \text{ mA/cm}^2$ , and 1000 h stability. The students will be involved in a multidisciplinary team to learn: catalysis, nanomaterials, 3D printing, and mass transport simulation by COMSOL.